Minimum Safety Requirements for Toxic, Corrosive and Reactive Compressed Gases



CBC Lab Safety Workshop
Beyond the Basics
Wednesday, March 16, 2011
University of Arizona Student Union
Tucson Room

Overview

- Define terms in presentation
- Provide specific gas examples
- Explain need for concern
- Explain general hazard control approach
- Introduce minimum requirements checklist
- Describe an application of requirements
- Campus resources

Toxic and Highly Toxic Gases <



- Toxic: LC₅₀ between 200 and 2000 parts per million *
- Highly toxic: LC₅₀ of less than 200 parts per million *

* when administered continuously by inhalation for one hour to albino rats

Corrosive Gases



 Gases that cause visible destruction or, or irreversible alterations in, living tissue by chemical action at the site of initial contact

Reactive Gases

 Unstable (reactive): Gases which in the pure state, or as produced or transported, will vigorously polymerize, decompose, condense, or will become self-reactive under conditions of shocks, pressure, or temperature

Compressed Gases

- Gases or mixture of gases having, in a container, an absolute pressure exceeding
 - 40 psi at 70°F (21.1°C); a gas or mixture of gases having, in a container, an absolute pressure exceeding OR
 - 104 psi at 130°F (54.4°C) regardless of the pressure at 70°F,
- or a liquid having a vapor pressure exceeding 40 psi at 100°F (37.8°C).

TOXIC, CORROSIVE AND REACTIVE COMPRESSED GASES

(examples)

Toxic and Highly Toxic

- carbon monoxide
- arsine
- chlorine
- cyanogen chloride
- diborane
- fluorine
- germane
- phosgene
- hydrogen cyanide
- hydrogen fluoride
- hydrogen selenide
- hydrogen sulfide
- nitric oxide
- nitrogen dioxide
- phosphine

Carcinogenic

- 1,3-butadiene
- vinyl chloride
- ethylene oxide



Corrosive

- ammonia
- chlorine
- hydrogen fluoride
- sulfur dioxide
- silicon tetrafluoride
- hydrogen chloride
- hydrogen bromide

Reactive

- silane
- germane
- dichloroborane
- phosphine
- stibine





- Difficult to contain
 - High vapor pressure (VP)
 - Under pressure
 - e.g., ammonia 114 psi @70°F
 - e.g., nitric oxide, chlorine, hydrogen cyanide 2000 psi
- Large quantity in a small package
 - e.g., 50" ammonia cylinder, 50 lbs liquid or 1039 ft³ gas

- Low exposure limit
 - e.g., arsine, hydrogen selenide 0.05 ppm (TWA)
 - e.g., cyanogen chloride 0.3 ppm (Ceiling)
 - e.g., fluorine 2 ppm (Ceiling)
 - e.g., ammonia, carbon monoxide 25 ppm (TWA)
 - contrast e.g., acetone 500 ppm (TWA)

- High vapor hazard ratio (VHR)
 - VHR = $(VP @ temperature of use)(10^6)/atmospheric pressure$ exposure limit
 - e.g., arsine $(114 \text{ psi})(10^6)/14.7 \text{ psi} = 155 \times 10^6$ 0.05
 - contrast e.g. acetone $(185 \text{ mm Hg})(10^6)/760 \text{ mm Hg} = 490$

- Some have history of serious incidents
 - e.g., silane







- 6 fatalities from gas cabinet explosions where silane was released unignited into the cabinet (two most recent - Taiwan 2005, India 2007,
- Other major incidents in China and U.S. in 2009

 Some are difficult to detect (i.e., poor warning properties)

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- e.g.,:
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- carbon monoxide
- sulfuryl fluoride
- hydrogen sulfide

Serious Hazards Require Serious Controls

- Specified Codes and Regulations
 - International Fire Code (IFC 2006)
 - Occupational Safety & Health Administration (OSHA)
- Accepted Consensus Standards/Industry Practice
 - Compressed Gas Association (CGA)
 - Semiconductor Equipment Manufacturers Institute (SEMI)
 - Gas suppliers
- Redundant controls if a single point failure could result in a significant accident or exposure above the occupational exposure limits (OELs)

Primary Hazard Controls

- Minimizing gas quantity
- Enclosing or isolating the source, distribution system and apparatus in which the gas is used

<u>Passive Hazard Controls</u> (require no power or human intervention after initial installation)

Ventilation

<u>Active Hazard Control</u> (rely on the presence of power or human intervention to be effective)

Passive Hazard Controls

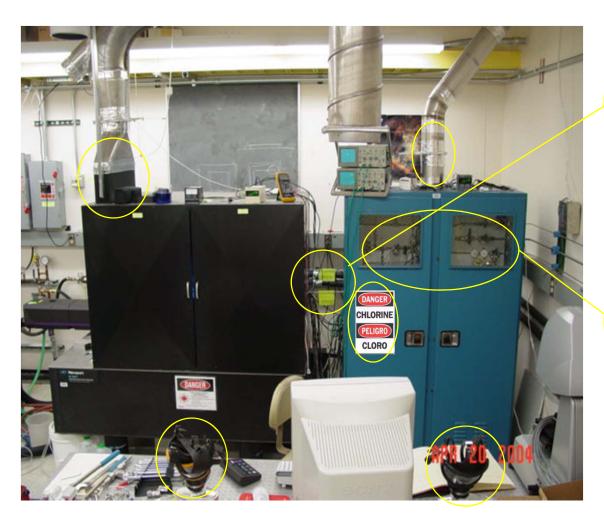
(examples)

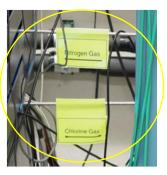
- Substitution with non-gaseous compounds
- Dilution in a compatible non-toxic, non-corrosive or non-reactive gas in the cylinder mixed by the vendor (i.e., reduced gas concentration)
- Onsite gas generation
- Quantity limits that can be stored at a given location and used within a reasonable time (i.e., reduced gas amount)
- Restrictive flow orifices (i.e., limited available gas)
- Use of all-welded gas delivery lines (i.e., reduced likelihood of gas leaks)

Active Hazard Controls

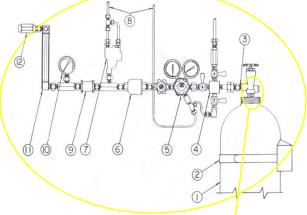
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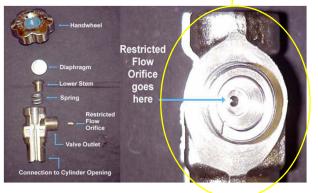
- Leak testing using an inert gas
- Ventilation with airflow monitor/alarm (regular testing and calibration)
- Operating at pressures below atmospheric
- Gas monitors (regular testing and calibration)
- Process shut-off devices











Minimum Safety Requirements Checklist

(http://risk.arizona.edu/healthandsafety/labchemicalsafety/MinRegforCSL-3ToxicCorrosiveGasUse.pdf)

- Quantity limits
- Personal protection and training
- Indoor cylinder location
- Piping, tubing, valves and fittings
- Equipment maintenance
- Environmental protection
- Label, signs and material safety data sheets
- Separation from incompatible materials
- Cylinder handling and transportation
- Security
- Hazard analysis













Carbon Monoxide

- Toxic and flammable compressed gas
- Inhibits the blood from carrying oxygen (brain and heart most susceptible to effects)
- No warning properties
- 76 ft³/ cylinder, 2000 psi
- Exposure limit = 25 ppm (TWA)
- VHR = 2.3×10^6































Chemical Safety Level 3 (CSL-3) Gases/Gas Activities

- The storage or use of compressed pyrophoric gases in flammable concentrations
- The storage or use of compressed highly toxic gases or compressed toxic gases with poor or no warning properties
- The storage or use of compressed gases, which are select carcinogens, reproductive toxins, toxic or corrosive:
 - in quantities greater than 20 ft³ OR
 - that are plumbed outside of the source exhausted enclosure
- Transfilling

Need Help?

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